

Chapter 12 Review Solutions

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Chapter 12 Review Solutions

Modern Chemistry 7 Solutions CHAPTER 12 REVIEW Solutions SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. Describe the errors made by the following students in making molar solutions. a. James needs a 0.600 M solution of KCl. He measures out 0.600 g of KCl and adds 1 L of water to the solid.

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CHAPTER 12 REVIEW Solutions SECTION
2 SHORT ANSWER Answer the following
questions in the space provided. 1. The
following are statements about the
dissolving process. Explain each one at
the molecular level. a. Increasing the
pressure of a solute gas above a liquid
solution increases the

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Chapter 12 Review Solutions. 1) 56 . 2) $= 4$, $= 10$, $= 16$, $= 22$, $= 28$. 3) $= 4$, $= 8$, $= 16$, $= 32$, $= 64$, $= 128$. 4) $an = 2n - 3$. 5) $= \cdot 4n$. 6) $= 5$, $= 3$, $= 1$, $= -1$. 7) $3 + 4 + 5 + \dots + (n + 2)$ 8)

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CHAPTER 12 REVIEW Solutions SECTION
1 SHORT ANSWER Answer the following
questions in the space provided. 1.

Match the type of mixture on the left to
its representative particle diameter on
the right. c solutions (a) larger than
1000 nm a suspensions (b) 1 nm to 1000
nm b colloids (c) smaller than 1 nm 2.

Identify the solvent in each of the
following examples:

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REVIEW SHORT ANSWER Answer the
following questions in the space
provided. 1. Solid CaCl_2 does not
conduct electricity. Explain why it is
considered to be an electrolyte. 2.
Explain the following statements at the
molecular level: a. Generally, a polar
liquid and a nonpolar liquid are
immiscible. b.

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Notes and Answers Chapter 12 SECTION
1 SHORT ANSWER 1. c 2. a 3. b 2. a.
alcohol b. water c. the gels 3. The
mixture is a colloid. Modern Chemistry
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Chapter 12 Solutions Review Answers

CHAPTER . 12 . REVIEW ""-- Solutions .
SHORT ANSWER Answer the following
questions in the space provided. 1.
Describe the errors made by the
following students in making molar
solutions. 3. James needs a 0.600 M
solution of KCl. He measures out 0.600 g
of KCl and adds 1 L of water to the solid.
James made several errors.

CHAPTER REVIEW Solutions

Modern Chemistry Chapter 12-.
Solutions. Section 1- Types of Mixtures
Solutions are homogeneous mixtures of
two or more substances in a single
phase. Soluble describes a substance as
capable of being dissolved. Solvent is
the dissolving medium in a solution.

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12 REVIEW Solutions Teacher Notes and

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Answers Chapter 12 SECTION 1 SHORT ANSWER 1 c 2 a 3 b 2 a alcohol b water c the gels 3 The mixture is a colloid The properties are consistent with those reported in CHAPTER 12 REVIEW Solutions -

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Overview

Modern Chemistry Chapter 12-Solutions
Section 1- Types of Mixtures Solutions
are homogeneous mixtures of two or
more substances in a single phase.
Soluble describes a substance as
capable of being dissolved. Solvent is
the dissolving medium in a solution.
Solute is the substance that is dissolved
in a solution.

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Review. A wire of length L , Young's modulus Y , and cross ...

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