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Study Guide. STUDY.  
PLAY. Anion. An ion  
with a negative charge.

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The charge for an Anion is written as a number followed by a minus sign. Cation. An ion with a positive charge. The charge for a cation is written as a number followed by a plus sign. Chemical formula.

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Chemistry Chapter 7 Study Guide. the force that holds two atoms together; may form by the attraction of a positive ion for a negative ion or by

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sharing electrons. the electrostatic force that holds oppositely charged particles together in an ionic compound.

### **Chemistry Chapter 7 Study Guide**

#### **Answers**

Chemistry Chapter 7.  
wave. crests. troughs.  
wavelength. a periodic  
back and forth motion  
that tranmits energy.  
the high points of a  
series of waves. the

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low points of a series of waves. the distance between any two corresponding points on successiv....

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Chapter 7 Study Guide: Chemistry. To find the number of valence electrons in an atom of a representative element, simply look at its group number. A



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positively charged ion, or cation, is produced when an atom loses one or more valence electrons. Most ionic compounds are crystalline solids at room temp.

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Sample Study Sheet

7.1: Balancing

Chemical

Equations describes the  
important skill of

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balancing equations so that they reflect the fact that the number of atoms of each element for the products of chemical changes always equals the number of atoms of each element in the initial reactants.

### **Chapter 7 An Introduction to Chemical Reactions**

Chapter 7 Energy and  
Chemical Reactions.

Review Skills 7.1

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Energy Kinetic Energy

Potential Energy Units

of Energy Kinetic

Energy and Heat

Radiant Energy 7.2

Chemical Changes and

Energy 7.3 Ozone:

Pollutant and Protector

Removal of UV

Radiation by Oxygen

and Ozone Molecules

The Natural

Destruction of Ozone

7.4

Chlorofluorocarbons: A

Chemical Success

Story Gone Wrong

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CFCs and the Ozone Layer.

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STUDY GUIDE FOR  
CHAPTERS 6 & 7

Certain topics like the structure, synthesis, and nomenclature of alkyl halides have already been covered previously. This part of the course is primarily concerned with  $S_N1$  and  $S_N2$  reactions,  $E1$  and  $E2$  reactions, and alkene synthesis, structure, and nomenclature.

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**STUDY GUIDE FOR  
CHAPTERS 6 & 7**

CHAPTER 7 REVIEW.

Chemical Formulas and  
Chemical Compounds.

SECTION 1. SHORT

ANSWERAnswer the  
following questions in  
the space provided. 1.

cln a Stock system  
name such as iron(III)  
sulfate, the Roman  
numeral tells us.

(a)how many atoms of  
Fe are in one formula  
unit.

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**7 Chemical Formulas  
and Chemical  
Compounds**

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Answer Key Modern  
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chemistry study guide  
answer key. Modern  
Chemistry : Chapter

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Tests with Answer Key

**Chapter 7 Chemistry  
Study Guide Answer  
Key**

and the Royal Society of Chemistry, first introduced a Study Guide for Qualified Persons in 1978 based on Article 23 of 75/319/EEC. Further revisions were completed in 2000, 2004, 2006 and 2008. This revision was made effective from 1



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Chemistry ...**

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Chemical Reactions 84

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Chemistry Section

Goals and

Introductions Section

71 Energy Goals To

introduce the terms  
energy, kinetic energy,  
and potential energy

To introduce the Law of

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Conservation of Energy  
To describe the  
relationships

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Change Chapter 16  
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Change Made for  
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2 Matter and Change

Matter and Change

part 1 ( Chem 1 H) Ch.

1 part 1 defines

chemistry, matter, and

physical and chemical

changes.

**Chemistry Matter**

**And Change Chapter**

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**Answers**

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Goals and

Introductions Section

71 Energy Goals To

introduce the terms

energy, kinetic energy,  
and potential energy

To introduce the Law of  
Conservation of Energy

To describe the  
relationships

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Chemistry is the study  
of matter: its  
composition,

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properties, and reactivity. This material roughly covers a first-year high school or college course, and a good understanding of algebra is helpful. Did you know that everything is made out of chemicals?

Chemistry is the study of matter: its composition, properties, and reactivity.

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