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(a) Determine the molar mass of CCl_2F_2 . From the balanced equation, calculate the moles of H_2 needed for the complete reaction. From the ideal gas law, convert moles of H_2 into volume. (b) $3.72 \times 10^3 \text{ L}$

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Answer Key Chapter

9 - Chemistry 2e |

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and Molecular

Structure: Orbital

Hybridization and

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Valence Bond Theory -

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Section 9-2 - Page 354

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Authors: Kotz, John C.;

Treichel, Paul M.;

Townsend, John R.;

Treichel, David A.,

ISBN-10: 1133949649

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**Chemistry and
Chemical Reactivity
(9th Edition)**

Chapter 9 ...

Chemistry is the study

of matter: its

composition,

properties, and

reactivity. This material

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roughly covers a first-year high school or college course, and a good understanding of algebra is helpful. Did you know that everything is made out of chemicals? Chemistry is the study of matter: its composition, properties, and reactivity.

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Notes - (Short Q,

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Chemistry Semester 2

Review Unit 9: The

Mole . What are the

following conversion

factors called? Include

the values moles A.)

moles & particles

Include the value of

any that are constants

8) grams & mole Cj

moles & liters (assume

STP) s called? 1 2.

Order the following

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quantities from
smallest to largest 05
mol D. 602 10 E. dozen
B. pair F. four moles ...

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Alcohol Syntheses 2 8

2-Carbon chain

extension Mech 9 1 0

Mech 10 Mech 11

NaBH₄ will not react

with estersH Mech

Review Routes to

Alcohols 10 M R O 11 1

, 12 13 N 1. OR'MgBr 2.

H₃O⁺ O ethyl n oxide

R' OH 1° alcohol HH HH

R'MgBr 1. 2. H₃O⁺ O R'

H 1° alcohol HH HH R H

R OH H °a lcoh 1.

LiAlH₄ aldehyde 2 ...

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**Review of Organic
Chem II - Minnesota
State University ...**

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a Khan Academy video.

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Periodic Table
Structure of the Atom
2. Quiz: Element
Symbols first 20
elements plus Br and I
transition metals: Sb,
As, Bi, Cd, Cr, Co, Cu,
Au, Fe, Pb, Mn, Hg, Ni,
Pt, Ag, Sn, Ti, Zn 3.

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Element symbol Rules
for writing the element
symbols: First letter
capitalize Second lower
case Hints ...

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below are not
necessarily in the order
that they will be
performed in the lab.
Please consult your
syllabus for the exact
schedule of your Chem

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Review And
9 experiments.

Measurements in the
Laboratory: Procedure
Report PPE: lab coat,
goggles, closed shoes.
Paper Chromatography
of Gel Ink Pens:
Procedure Report

Chem 9 Experiments

$2\text{N}_2(\text{g}) \rightarrow \text{NO}_2(\text{g})$ a.

Balance the equation.

$2\text{N}_2(\text{g}) + 3\text{O}_2(\text{g}) \rightarrow$

$4\text{NO}_2(\text{g})$ 4 mol NO_2 2:3

mol O_2 b. What is the
mole ratio of NO_2 to O_2 ?

15.0 mol c. If 20.0

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mol of NO_2 form, how many moles of O_2 must have been consumed? True d.

Twice as many moles of NO_2 form as moles of N_2O are consumed.

True or False? False e.

Twice as many grams of NO_2 ...

mc06se cFMsr i-vi

GRADE 9 CHEMISTRY

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Electricity. YouTube

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YouTube Video. Fourth

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Review And
week of notes.

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YouTube Video.

YouTube Video.

YouTube Video. Fifth
week of notes.

Electricity Calculations
powerpoint. Sixth week
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student will be

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introduced to the core concepts of chemistry

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Chemistry 11 Course

Review. Chemistry

11—Course Review

Page 19. 77oC. 43oC.

15oC. 90oC.

Temperature oC. 0.0

5.0 15.0 20.0 28.0

Time (minutes) A + B.

Energy (kJ) Reaction

Proceeding (AB.

Sheet3. Sheet2.

Sheet1. Chart1

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