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created nor destroyed during ordinary chemical reactions or physical changes $2H_2 + O_2 = 2H_2O$. law of definite proportions.

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Name _____ Date _____ Class _____ FILL IN THE BLANK Write the correct term (or terms) in the space provided. 9. If a particular compound is composed of elements A and B, the ratio of the mass of B to the mass of A will always be the same. This is a statement of the law of definite proportions. 10. The atomic mass of carbon-12 is referred to as a(n) 11.

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