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CHAPTER 5 REVIEW The Periodic Law SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. c In the modern periodic table, elements are ordered (a) according to decreasing atomic mass. (b) according to Mendeleev's original design. (c) according to increasing atomic number. (d) based on when they were discovered. 2. d Mendeleev noticed that certain similarities in the ...

5 The Periodic Law

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3. What orbitals combine together to form sp3 hybrid orbitals around a carbon atom? the s orbital and all three p orbitals from the second energy level 4. What two factors determine whether or not a molecule is polar? electronegativity difference and molecular geometry or unshared electron pairs 5.

6 Chemical Bonding

Section Quiz: The Development of a New Atom! In the space provided, write the letter of the term or phrase that best completes ... Modern Chemistry 19 . Name Section Quiz, continued Class Date ectra of atoms with how many 6. Bohr's model corr ctly explains the electron(s)? a one . two

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